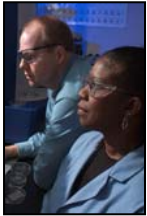


Impact of Alkali Source on Vitrification of SRS High Level Waste



We Put Science To Work

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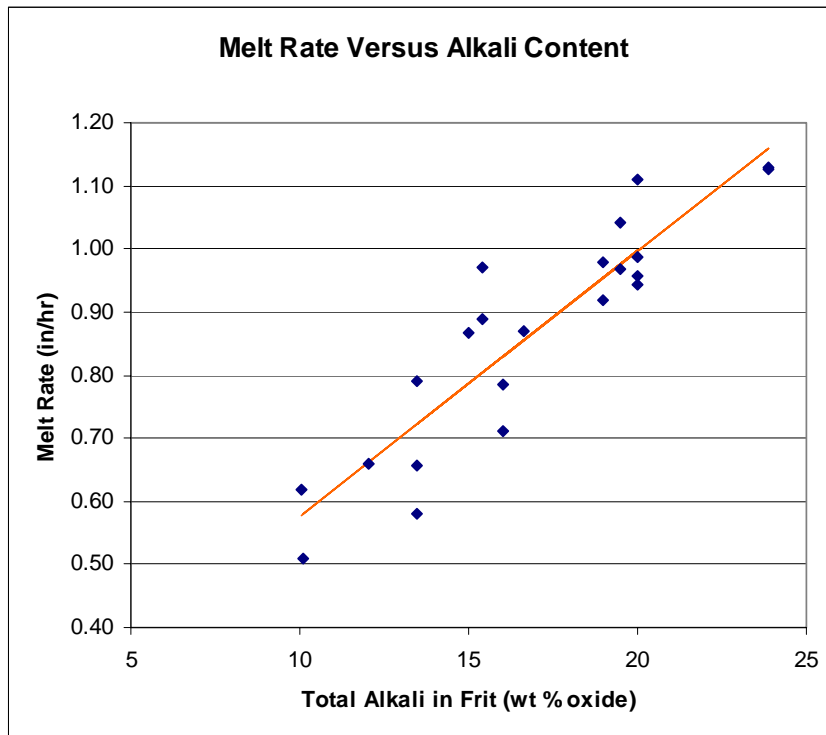
Overview

- Introduction
- Objectives
- Experimental Methods
- Results
- Conclusions
- Questions

Introduction

- The Savannah River Site (SRS) is currently immobilizing high level radioactive waste slurries from Cold War operations at the Defense Waste Processing Facility (DWPF)
 - Insoluble species are metal oxides/hydroxides
 - Fe, Al, Mn, U, Ni, Ca, Mg
 - Soluble species are primarily sodium salts
 - Nitrite, nitrate, hydroxide, carbonate
 - Vitrification used to stabilize waste
- Production rate improvement for DWPF decreases costs and speeds stabilization

Introduction: Alkali Impacts on Vitrification Plant



- Alkali content is a major factor in the operation of DWPF
 - Vitrification Impacts
 - Melt Rate
 - Offgas Quantity and Composition
 - Glass Quality Impacts
 - Durability
 - Glass pool viscosity
 - Pretreatment Impacts
 - Acid addition amount to acidify sludge
 - Rheological properties of process slurries
 - Offgas Quantity and Composition

Introduction

- Waste alkali
 - Sodium and potassium salts
 - Sludge washing strategy adjustable
 - Sludge is washed in “sludge batches”
 - Each batch represents 2 to 3 years of DWPF processing
 - SB3 washed to ~ 1M soluble sodium
- Frit alkali
 - Sodium and lithium oxides
 - Frit composition adjustable to reach “optimal” composition for a given sludge batch
- Emphasis during source of alkali testing was on sodium content

Objective

- Evaluate how the source of alkali (waste versus frit) impacts the pretreatment and vitrification processes of the Defense Waste Processing Facility.

Experimental Methods

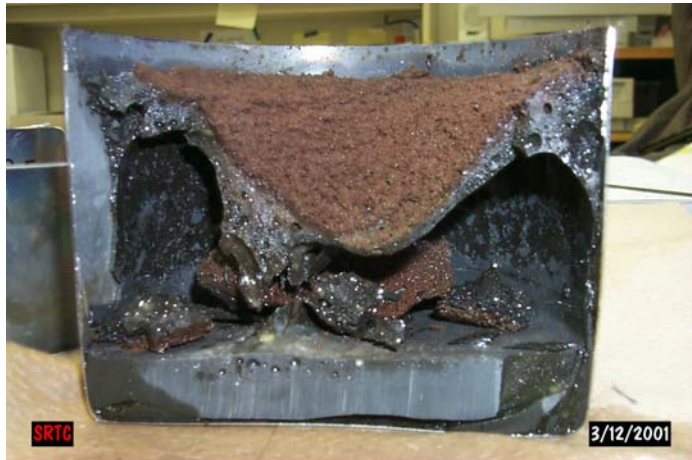
- Non-radioactive simulants used during testing
- Five simulants prepared to simulate varying the sludge washing
 - Sludge Batch 3 used as midpoint
 - Simulants named for amount of sodium in corresponding frit
 - 0% is least washed, 16% is most washed
- Five frits prepared to match the five simulants
 - Frit 418 used for midpoint (8% Na)
 - Frits named for amount of sodium in frit
- Melter feeds prepared targeting the same final glass composition
 - Waste loading adjusted as required to maintain constant final glass composition

Experimental Methods: Pretreatment



- Small-scale (2.5 liter) tests conducted to replicate the first stage of pretreatment (SRAT cycle)
 - Frit addition not performed (SME cycle)
 - Acid addition factor was set to 155% for all runs
 - Offgas composition measured with an on-line gas chromatograph
 - Samples taken for composition and physical property measurements
- Larger scale (15L) runs performed to provide feed for slurry-fed melt rate testing
 - SRAT/SME cycle performed

Experimental Methods: Melt Rate



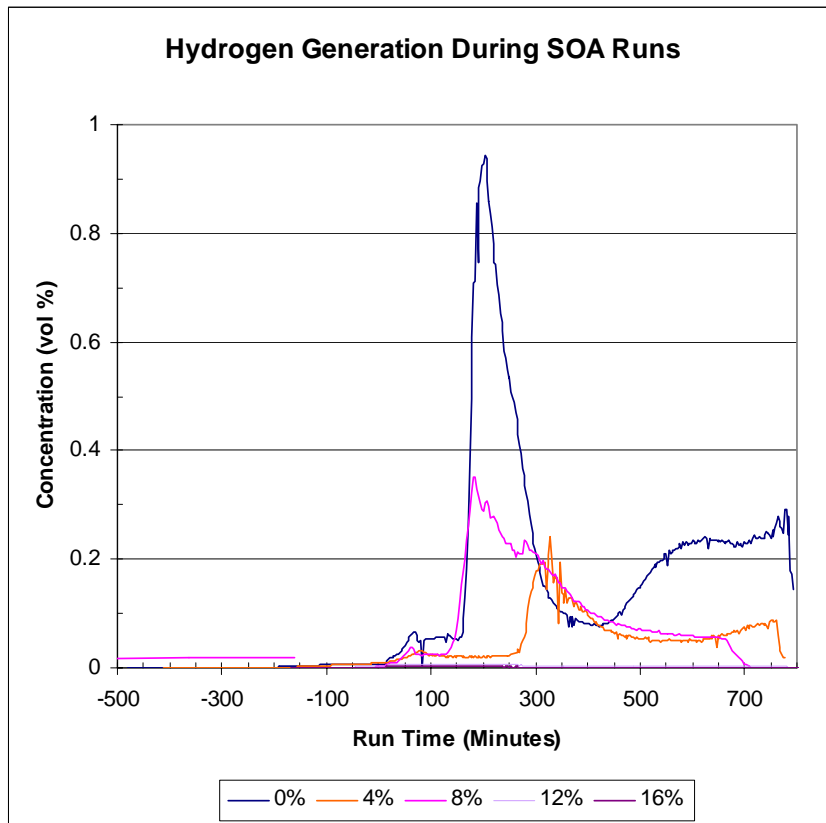
SB3: Frit 200



SB3: Frit 320

- Dry-Fed Melt Rate Furnace (MRF)
 - 4" diameter beaker
 - Dried feed added to beaker
 - Beaker sides/insulated
 - Beaker placed in top loading furnace for ~ 50 minutes
 - Beaker pulled and sectioned
 - Glass amount measured
- Slurry-Fed Melt Rate Furnace (SMRF)
 - 8" diameter test vessel
 - Slurry-fed in increments of ~ 90 ml
 - Each increment added when vapor space reaches setpoint
 - Continuous overflow glass pouring
 - Melt rate determined from glass pour rate

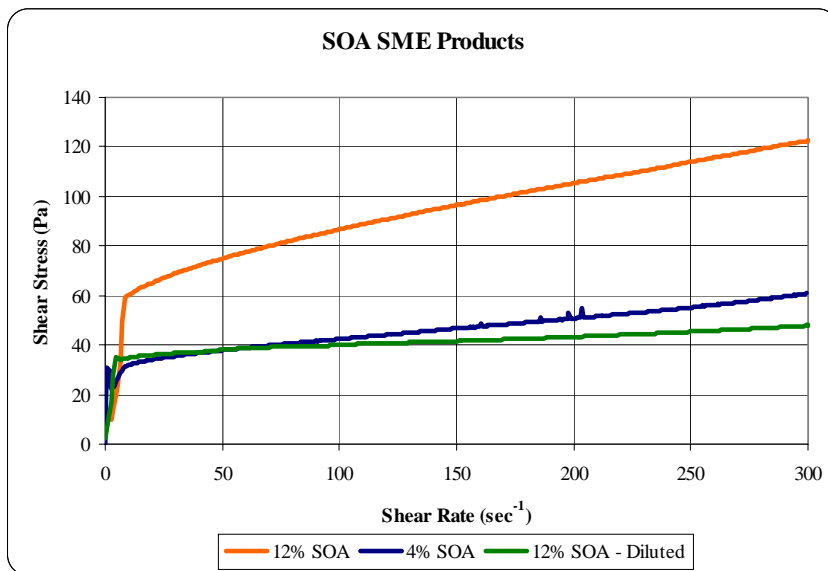
Results: Pretreatment Offgas Composition



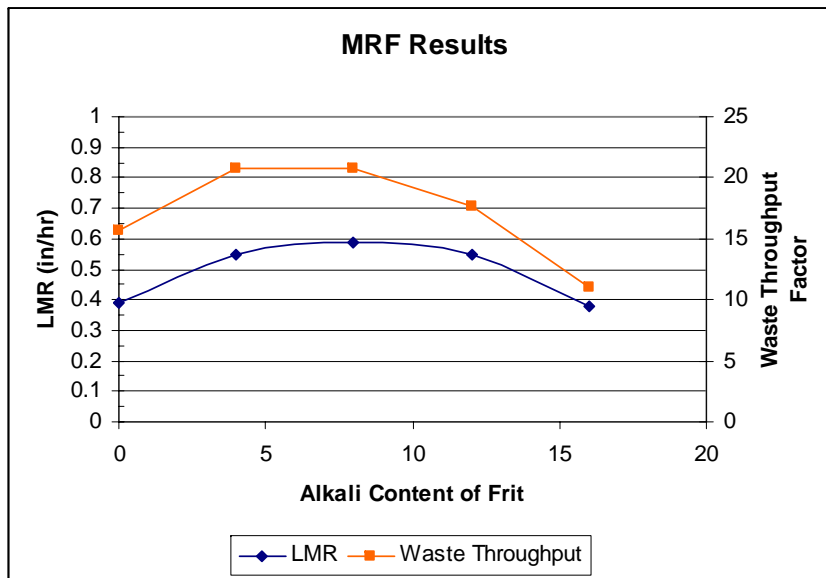
- Higher levels of offgas generation noted in less washed feed
 - Some soluble species (nitrite, carbonate, etc) generate gases
 - Larger acid demand adds more formic acid to generate hydrogen
 - Hydrogen generation significantly greater for less washed feeds

Results: Rheology of Melter Feed

- Melter feed rheology impacted by amount of sodium in sludge feed
 - Same targeted solids (50% total solids) results in higher insoluble solids for more washed feed
 - Adjusting more washed feed to same insoluble solids concentration as less washed feed yields similar yield stress values



Results: Melt Rate



- MRF results indicated a broad optimal region
 - Melt rate maximum centered on SB3 baseline
 - 21% Na in sludge (oxide basis)
 - 8% Na in frit (8% Li also)
 - Waste throughput factor optimal to the less washed side of SB3 baseline
 - Refractory frit or waste on either side of optimal point likely hindered melt rate
- SMRF results confirmed results of MRF testing
 - More washed melter feeds required dilution prior to testing

Conclusions

- Broad optimal region for sodium content in the waste in terms of melt rate impacts
- Less washed waste requires more acid during pretreatment
- Less washed waste generates more offgas (including hydrogen) during pretreatment and vitrification
- Less washed waste is less viscous at same total solids content due to higher soluble solids content

Questions
